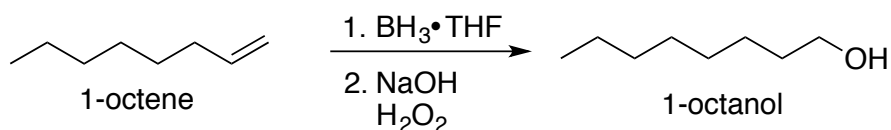


Chemistry 233

Infrared Spectroscopy Spring 2018

Consider Lab 9 – Hydroboration/Oxidation:

How do we know that the reaction has occurred as expected?



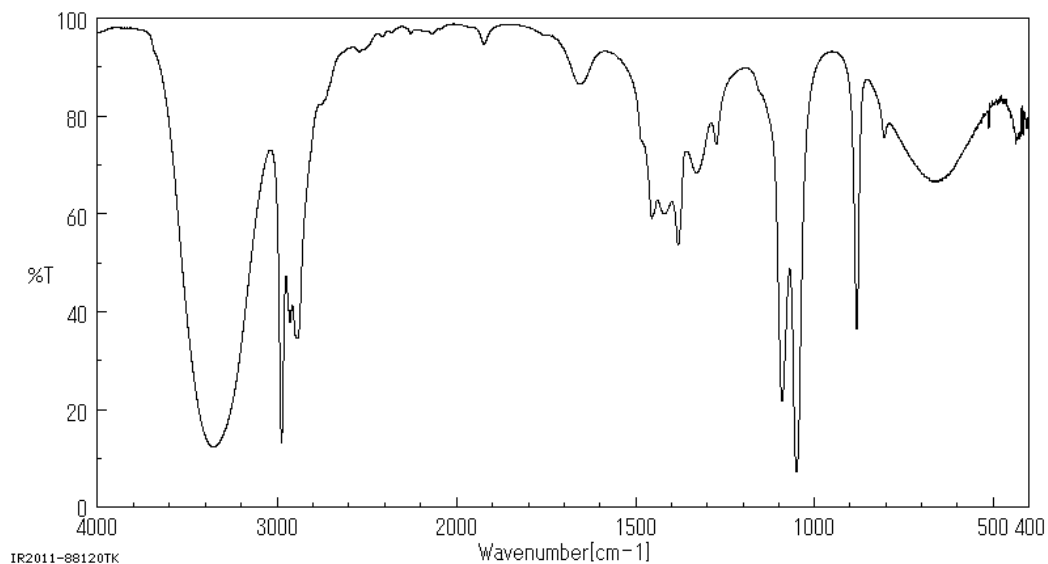
1

Infrared Spectroscopy

- **Main Purpose = determination of functional groups in a molecule.**
- **Generally can not deduce an entire structure from IR.**
- **When IR light is passed through a molecule, the bonds will absorb IR light causing the bonds to stretch or bend.**
- **This energy absorption can be seen on the IR spectrum.**

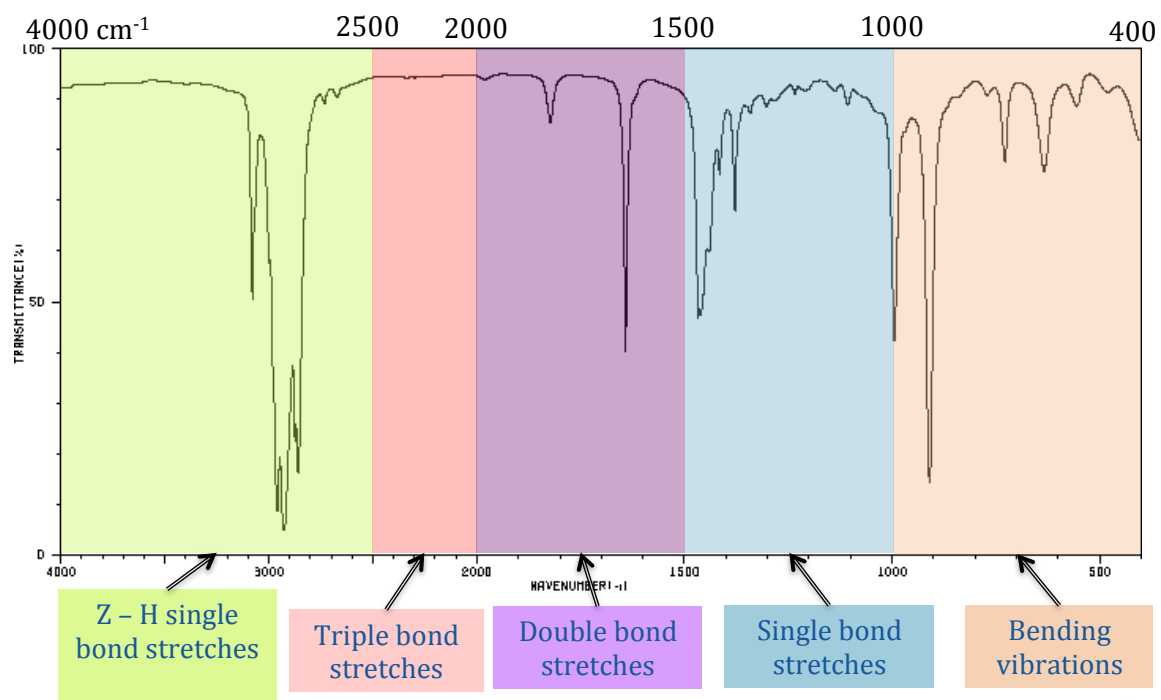
2

Sample IR Spectrum



3

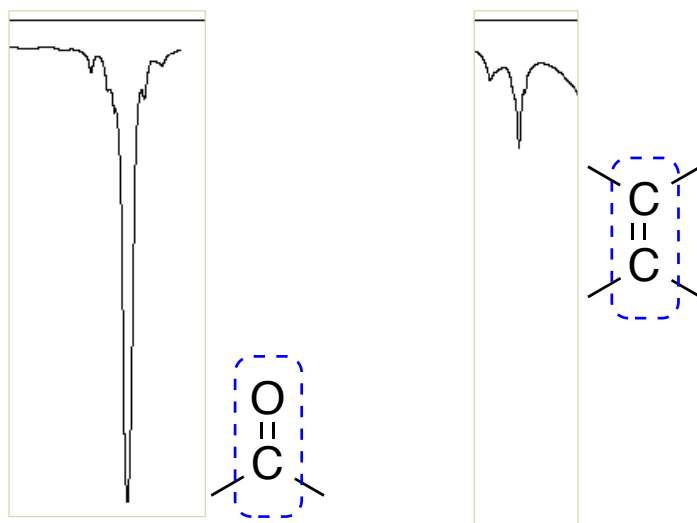
General IR Regions



4

Signal Intensity

A more polar bond will have a stronger absorption than a less polar bond.

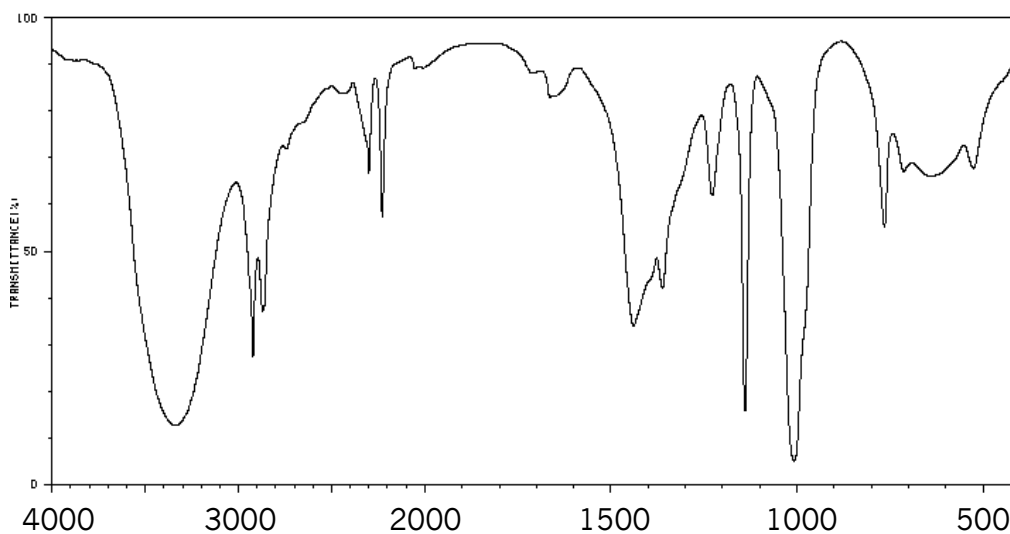


5

Key Functional Groups to Look For

1) Alcohol O-H Stretch

Broad Stretch, 3200-3600 cm^{-1}

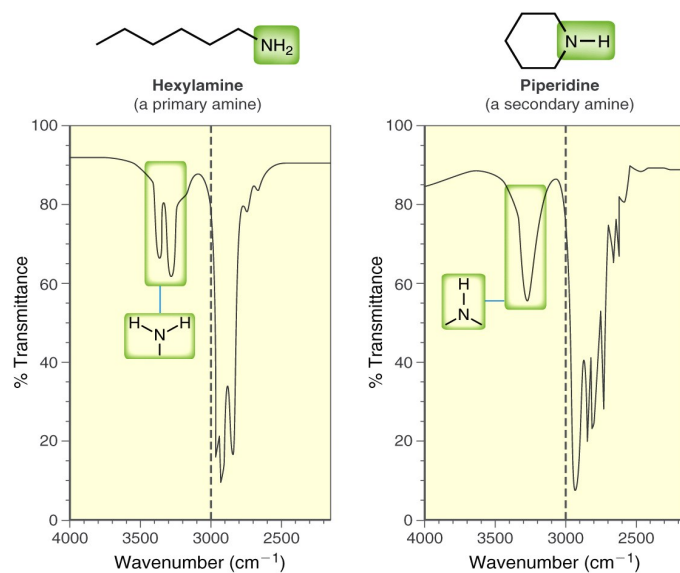


6

Key Functional Groups to Look For

2) Amine/Amide N-H Stretch

Medium Stretch, 3300-3500 cm^{-1}

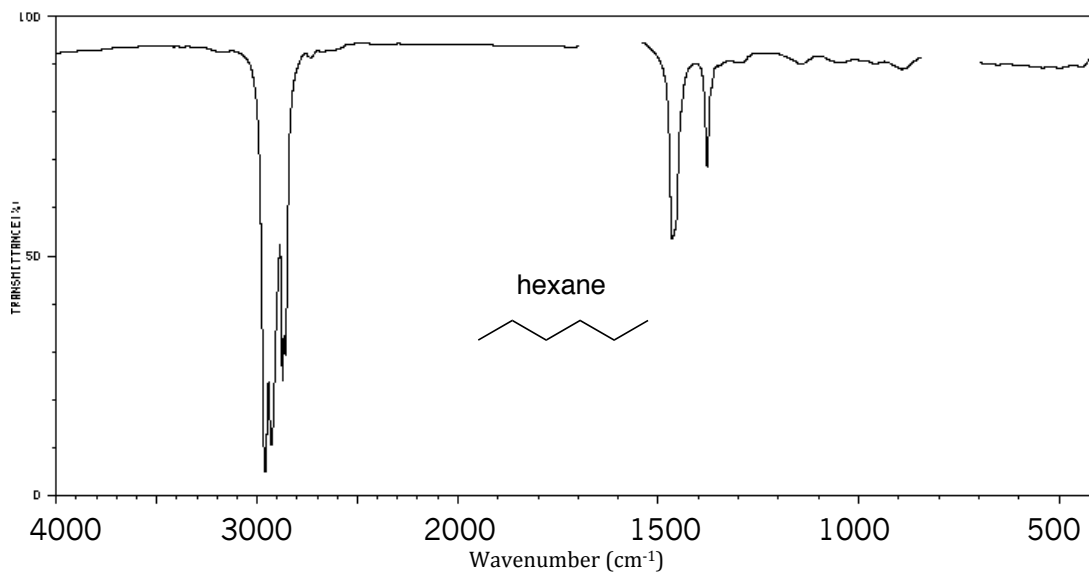


7

Key Functional Groups to Look For

3) Alkane - $\text{Csp}^3\text{-H}$ Stretch

Stretches just below 3000 cm^{-1}

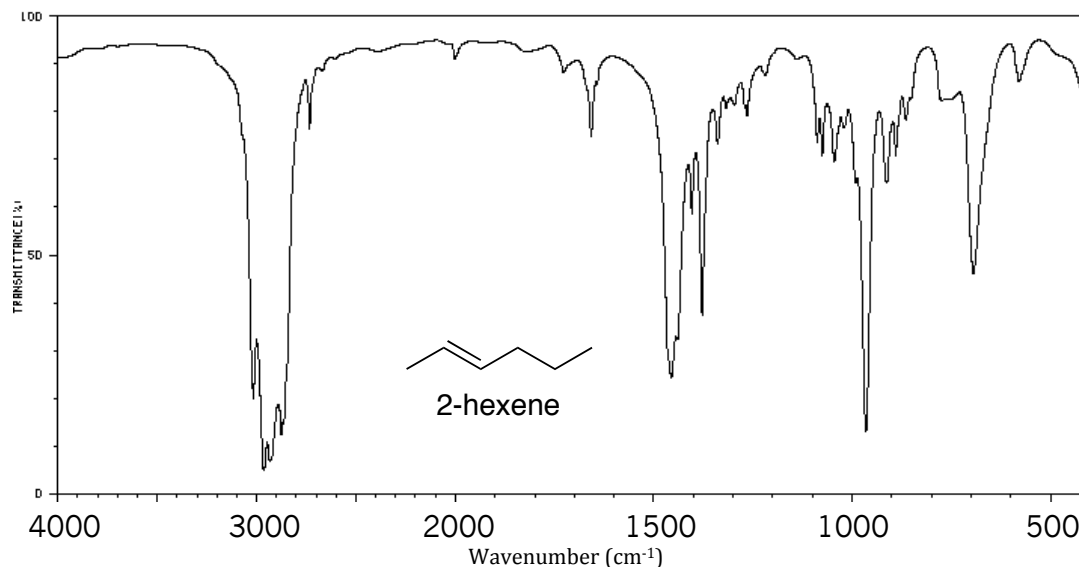


8

Key Functional Groups to Look For

4) Alkene *Csp²-H just above 3000 cm⁻¹*

C=C stretch 1600-1700 cm⁻¹

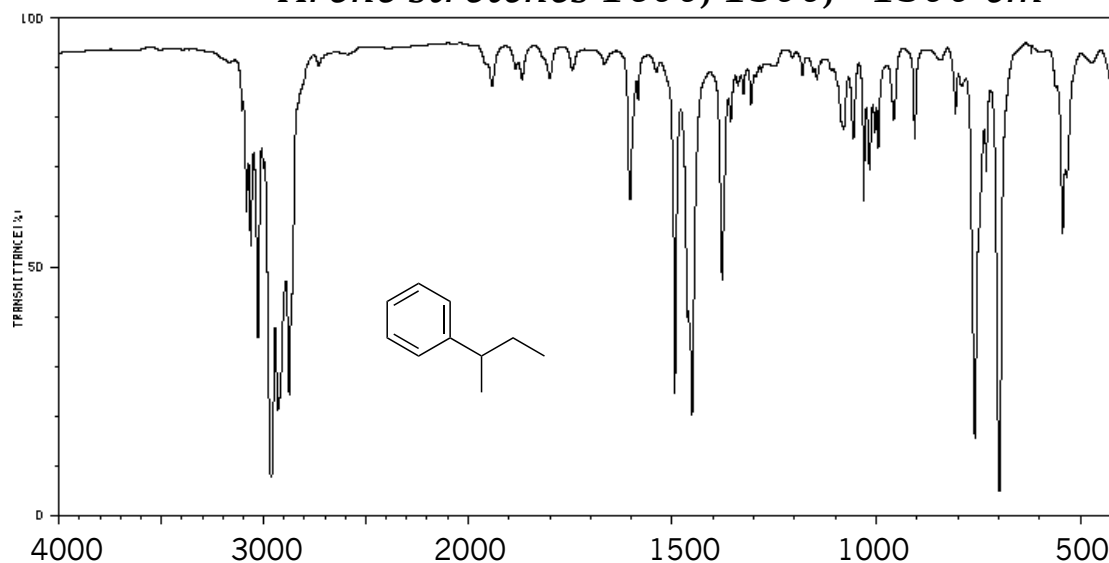


9

Key Functional Groups to Look For

5) Arene *Csp²-H just above 3000 cm⁻¹*

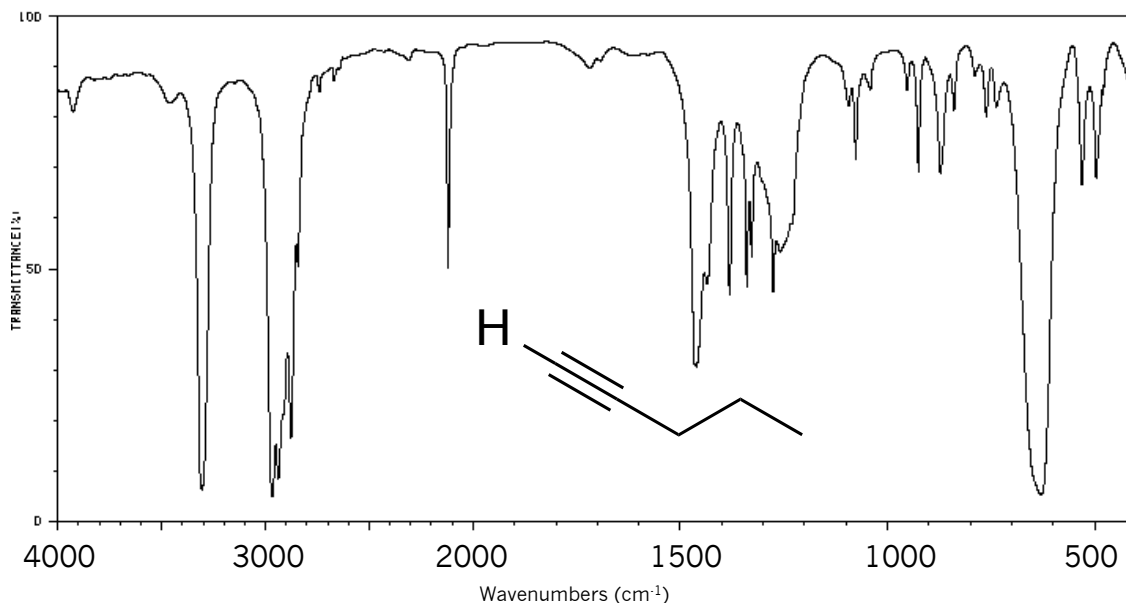
Arene stretches 1600, 1500, <1500 cm⁻¹



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Key Functional Groups to Look For

6) Alkyne $C_{sp}-H \sim 3300\text{ cm}^{-1}$ (terminal alkyne)
 $C\equiv C$ stretch $\sim 2150\text{ cm}^{-1}$



Key Functional Groups to Look For

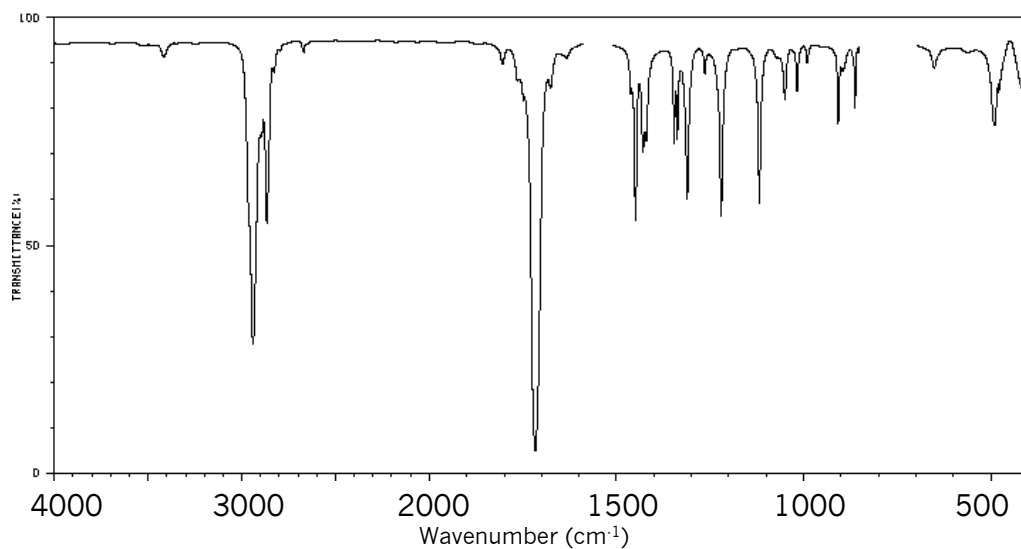
7) Carbonyl Groups

Look for the $C=O$ functional group in the $1650 - 1750\text{ cm}^{-1}$ region.

Use other features to distinguish the various carbonyl groups from one another.

A Ketone

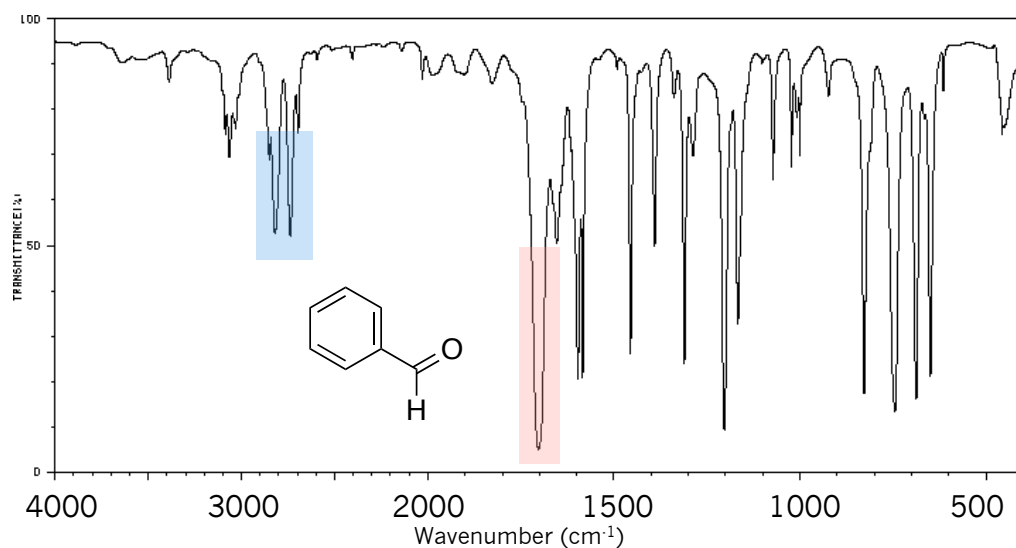
Strong sharp stretch $\sim 1700\text{ cm}^{-1}$



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An Aldehyde

In aldehydes, look for specific aldehyde C-H stretches at ~ 2720 and $\sim 2820\text{ cm}^{-1}$.

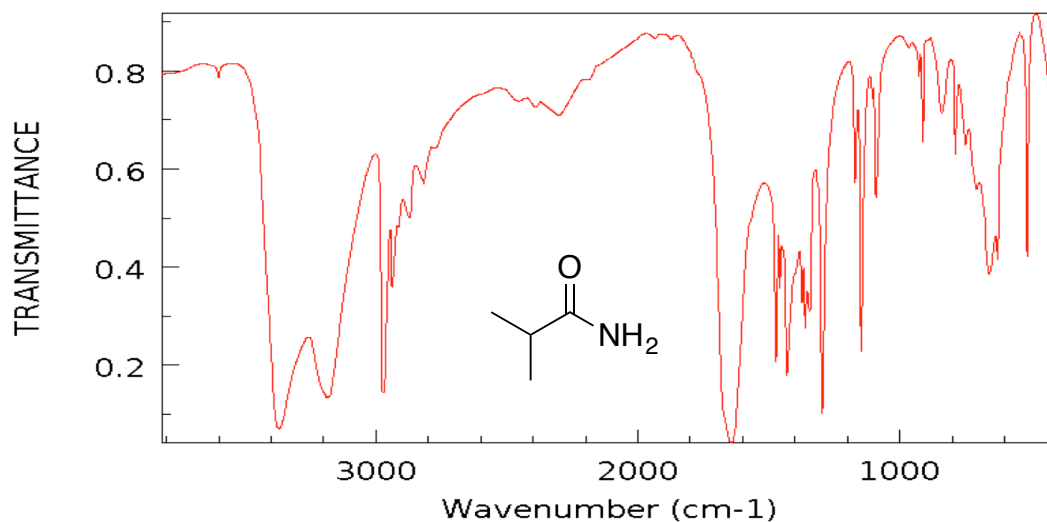


14

An Amide

Look for the C=O stretch $\sim 1650\text{ cm}^{-1}$

Also look for N-H stretches above 3000 cm^{-1}



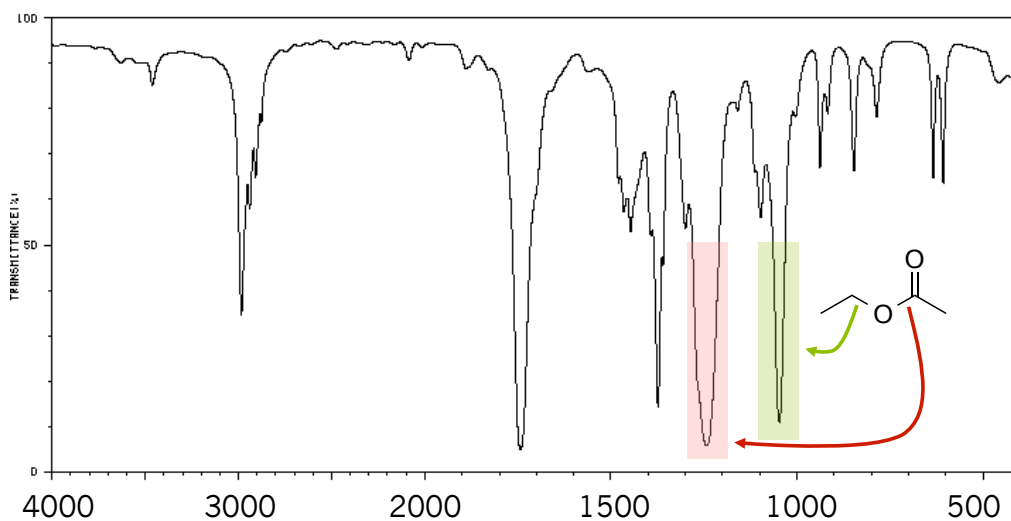
15

An Ester

Look for the C=O stretch $\sim 1750\text{ cm}^{-1}$

Csp³-O stretch: $1000\text{--}1100\text{ cm}^{-1}$

Csp²-O stretch: $1200\text{--}1300\text{ cm}^{-1}$



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Hydroboration Reaction

